



**TEACHING INSTRUCTIONAL DESIGN (BRP)
COURSE
INSTRUMENTATION PHYSICS 2**

by

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PREFACE

The Teaching Instructional Design (BRP) contains lesson plans for one semester. BRP was prepared to be used as a reference for learning the Instrumentation Physics 2 course at the Department of Physics, FMIPA UI.

The Instrumentation Physics 2 course is scheduled to be followed by 6th semester physics students, with the requirement that the student has taken the Physical Instrumentation 1 course. In this course, students will learn the instrumentation methods and techniques that are widely used in Physics. At the end of the lecture, students will present the assignment given by the lecturer in accordance with the lecture topic.

With the preparation of this BRP, it is hoped that it can become a reference for the learning process for lecturers and for students participating in lectures in particular and for people who want to learn it.

Depok, May 2016

Dr. Cuk Imawan

I. Informasi Umum

1. Name of Program / Study Level : Physics / Undergraduate
2. Course Name : Instrumentation Physics 2
3. Course Code : SCFI603714
4. Semester : 5
5. Credit : 2 Credits
6. Teaching Method(s) : Collaborative Learning
7. Prerequisite course(s) : Instrumentation Physics 1
8. Requisite for course(s) : Instrumentation specialization courses
9. Integration Between Other Courses : -
10. Lecturer : Dr. Cuk Imawan
11. Course Description : To describe the instrumentation methods and techniques that are widely used in Physics.

II. Course Learning Outcome (CLO) and Sub-CLOs

A. CLO

Students are able to explain basic concepts related to instrumentation based on physics and mathematics (C3) (ELO 3, 5, 6, 7)

B. Sub-CLOs

1	Applying Learning Methods (C2)	1
2	Describing Simple Measurements (C2)	1
3	Describing Operational Amplifiers and Their Applications (C2)	1
4	Describe Digital and Computer Electronics (C2)	1
5	Describe Signals and Noise (C2)	1
6	Describe the Introduction to the Spectrometric Method (C2)	1
7	Describing the Components of Optical Instrumentation (C2)	2
8	Describing Introduction to the Optical Atomic Spectrometry (C2)	2
9	Describing Atomic Absorption and Atomic Fluorescence Spectrometry (C2)	1
10	Describing Atomic Emission Spectrometry (C2)	1
11	Describing Atomic Mass Spectrometry (C2)	1
12	Describing Atomic X-ray Spectrometry (C2)	1

III. Rencana Pembelajaran

Week	Sub-CLO	Study Materials	Teaching Method	Time Required	Learning Experiences (*O-E-F)	Sub-CLO Weight on Course (%)	Sub-CLO Achievement Indicator	References
1	1	Introduction	Collaborative Learning	100 minutes	O:(70%) E:(0%) F:(30%)	4	Explain Learning Methods	
2	2	Simple Measurement: Electronic Components and Circuits	Collaborative Learning	100 minutes	O: (20%) E: (50%) F: (30%)	7	Explain the basic concepts: a) DC circuit b) AC circuit c) Semiconductors and semiconductor devices d) Ppower supply and regulator e) Reading Device	Sec. 1 Page. 26
3	3	Operational Amplifiers in Chemical Instrumentation	Collaborative Learning	100 minutes	O: (20%) E: (50%) F: (30%)	7	Explain the basic concepts: a) Properties of the Op-Amp b) Op-Amp circuit c) Transducer signal amplification and measurement d) Op-Amp applications for voltage and current control e) Op-Amp Application for Mathematical Operations	Sec. 1 Page . 59
4	4	Electronics Digital and Computer	Collaborative Learning	100 minutes	O: (20%) E: (50%) F: (30%)	7	Explain the basic concepts: a) Analog and Digital Signals b) <i>Counting and Arithmetic with Binary Numbers</i>	Sec. 1 Page . 80

							<ul style="list-style-type: none"> c) Basic Digital Circuits d) Computers and computerized instruments e) Computer Components f) Computer Software g) Computer Applications h) Computer Networks 	
5	5	Signal and Noise	Collaborative Learning	100 minutes	O: (20%) E: (50%) F: (30%)	7	<p>Explain the basic concepts:</p> <ul style="list-style-type: none"> a) Signal to noise Ratio (SNR) b) Sources of noise in Instrumentation Analysis c) Signal-to-Noise enhancements 	Sec. 1 Page . 110
6	6	Introduction to the Spectrometric Method	Collaborative Learning	100 minutes	O: (20%) E: (50%) F: (30%)	7	<p>Explain the basic concepts:</p> <ul style="list-style-type: none"> a) general properties of electromagnetic radiation b) the nature of electromagnetic radiation waves c) quantum mechanical properties of radiation d) The quantitative aspect of spectrochemical measurements 	Sec. 2 Page . 132
7	7	Optical Instrumentation Components	Collaborative Learning	100 minutes	O: (20%) E: (50%) F: (30%)	8	<p>Explain the basic concepts:</p> <ul style="list-style-type: none"> a) General design of optical instruments b) Radiation Sources c) Wavelength selector d) Sample Containers 	Sec. 2 Page . 164

							e) Radiation transducer	
8	7	Optical Instrumentation Components	Collaborative Learning	100 minutes	O: (20%) E: (50%) F: (30%)	8	Explain the basic concepts: a) Signal Processing and Readout b) Fiber Optic c) Types of Optical Instruments d) Fourier Transform Principle in Optical Measurement	Sec. 2 Page . 164
9	Mid Term Exam							
10	8	Introduction to Optical Atomic Spectrometry	Collaborative Learning	100 minutes	O: (20%) E: (50%) F: (30%)	7	Explain the basic concepts: a) Optical atomic spectra b) Atomization Method	Sec. 2 Page . 215
11	8	Introduction to Optical Atomic Spectrometry	Collaborative Learning	100 minutes	O: (20%) E: (50%) F: (30%)	7	Explain the basic concepts: a) Introduction to Sample Methods	Sec. 2 Page . 215
12	9	Atomic Absorption and Atomic Fluorescence Spectrometry	Collaborative Learning	100 minutes	O: (20%) E: (50%) F: (30%)	7	Explain the basic concepts: a) Sample atomization technique b) Atomic Absorption Instrumentation c) Interference in Atomic Absorption Spectroscopy d) Atomic Absorption Analytical Techniques e) Atomic fluorescence spectroscopy	Sec. 2 Page . 230
13	10	Atomic Emission Spectrometry	Collaborative Learning	100 minutes	O: (20%) E: (50%) F: (30%)	8	Explain the basic concepts: a) Emission spectroscopy based on plasma source	Sec. 2 Page . 254

							b) emission spectroscopy based on Arc and Spark sources c) Another source for optical emission spectroscopy	
14	11	Atomic Mass Spectrometry	Collaborative Learning	100 minutes	O: (20%) E: (50%) F: (30%)	8	Explain the basic concepts: a) Some general features of atomic mass spectrometry b) Mass spectroscopy c) Inductively coupled plasma <i>Mass Spectrometry</i> d) Source of Mass Spectrometry Spark e) <i>Glow discharge Mass Spectrometry</i> f) Other Mass Spectrometric Methods	Sec. 2 Page . 281
15	12	Atomic X-ray Spectrometry	Collaborative Learning	100 minutes	O: (20%) E: (50%) F: (30%)	8	Explain the basic concepts: a) Basic Principles b) Instrument Components c) X-Ray Fluorescence Method d) X-Ray Absorption Method e) Electron microprobe	Sec. 2 Page . 303
16	Final Exam							

*) O : Orientation
 E : Exercise
 F : Feedback

Reference:

1. Skoog, Douglas A, *Principles of Instrumental Analysis, 6th Ed*, Thompson Higher Education – Canada, 2007

IV. Assignment Design

Week	Assignment Name	Sub-CLOs	Assignment	Scope	Working Procedure	Deadline	Outcome
1	Individual Assignments & Group Assignments	1	Question	Learning methods	Group, Independent and Online	100 minutes	-
2	Individual Assignments & Group Assignments	2	Question	a) DC circuit b) AC circuit c) Semiconductors and semiconductor devices d) Ppower supply and regulator e) Reading Device	Group, Independent and Online	100 minutes	Student power points, presentation results, independent assignment sheets
3	Individual Assignments & Group Assignments	3	Question	a) Properties of the Op-Amp b) Op-Amp circuit c) Transducer signal amplification and measurement d) Op-Amp applications for voltage and current control e) Op-Amp Application for Mathematical Operations	Group, Independent and Online	100 minutes	Student power points, presentation results, independent assignment sheets
4	Individual Assignments & Group Assignments	4	Question	a) Analog and Digital Signals b) <i>Counting and Arithmetic with Binary Numbers</i> c) Basic Digital Circuits d) Computers and computerized instruments e) Computer Components f) Computer Software g) Computer Applications h) Computer Networks	Group, Independent and Online	100 minutes	Student power points, presentation results, independent assignment sheets
5	Individual Assignments & Group Assignments	5	Question	a) Signal to noise Ratio (SNR) b) Sources of noise in Instrumentation Analysis c) Signal-to-Noise enhancements	Group, Independent and Online	100 minutes	Student power points, presentation results, independent assignment sheets
6	Individual Assignments & Group Assignments	6	Question	a) general properties of electromagnetic radiation	Group, Independent and Online	100 minutes	Student power points, presentation results,

				<ul style="list-style-type: none"> b) the nature of electromagnetic radiation waves c) quantum mechanical properties of radiation d) The quantitative aspect of spectrochemical measurements 			independent assignment sheets
7	Individual Assignments & Group Assignments	7	Question	<ul style="list-style-type: none"> a) General design of optical instruments b) Radiation Sources c) Wavelength selector d) Sample Containers e) Radiation transducer 	Group, Independent and Online	100 minutes	Student power points, presentation results, independent assignment sheets
8	Individual Assignments & Group Assignments	7	Question	<ul style="list-style-type: none"> a) Signal Processing and Readout b) Fiber Optic c) Types of Optical Instruments d) Fourier Transform Principle in Optical Measurement 	Group, Independent and Online	100 minutes	Student power points, presentation results, independent assignment sheets
10	Individual Assignments & Group Assignments	8	Question	<ul style="list-style-type: none"> a) Optical atomic spectra b) Atomization Method 	Group, Independent and Online	100 minutes	Student power points, presentation results, independent assignment sheets
11	Individual Assignments & Group Assignments	8	Question	<ul style="list-style-type: none"> a) Introduction to Sample Methods 	Group, Independent and Online	100 minutes	Student power points, presentation results, independent assignment sheets
12	Individual Assignments & Group Assignments	9	Question	<ul style="list-style-type: none"> a) Sample atomization technique b) Atomic Absorption Instrumentation c) Interference in Atomic Absorption Spectroscopy d) Atomic Absorption Analytical Techniques e) Atomic fluorescence spectroscopy 	Group, Independent and Online	100 minutes	Student power points, presentation results, independent assignment sheets
13	Individual Assignments & Group Assignments	10	Question	<ul style="list-style-type: none"> a) Emission spectroscopy based on plasma source b) emission spectroscopy based on Arc and Spark sources c) Another source for optical emission spectroscopy 	Group, Independent and Online	100 minutes	Student power points, presentation results, independent assignment sheets

14	Individual Assignments & Group Assignments	11	Question	<ul style="list-style-type: none"> a) Some general features of atomic mass spectrometry b) Mass spectroscopy c) Inductively coupled plasma <i>Mass Spectrometry</i> d) Source of Mass Spectrometry Spark e) <i>Glow discharge Mass Spectrometry</i> f) Other Mass Spectrometric Methods 	Group, Independent and Online	100 minutes	Student power points, presentation results, independent assignment sheets
15	Individual Assignments & Group Assignments	12	Question	<ul style="list-style-type: none"> a) Basic Principles b) Instrument Components c) X-Ray Fluorescence Method d) X-Ray Absorption Method e) Electron microprobe 	Group, Independent and Online	100 minutes	Student power points, presentation results, independent assignment sheets

V. Assessment Criteria (Learning Outcome Evaluation)

Evaluation Type	Sub-CLOs	Assessment Type	Frequency	Evaluation Weight (%)
Online Activeness	2 - 12	Activities at Scale	2 x 3	10
Final Paper / Presentation: 1. HG group discussion 2. FG group discussion 3. HG group discussion (verification)	2 – 7 and 8 – 12	Assessment sheet	2 x 3	10
Topic Presentation (HG)	2 – 7 or 8 – 12	PowerPoint	1	20
Individual Assignment	2 - 12	Assignment sheet	3	10
Group Assignment	2 - 12	PowerPoint	3	10
Exam 1	2 - 7	Essay Exam Questions	1	20
Exam 2	8 - 12	Essay Exam Questions	1	20
Total				100

VI. Rubric(s)

A. Criteria of Presentation Score

Score	Delivering the Presentation
85-90	Groups are able to convey explanations logically, fluently, and on time and are able to answer questions from fellow students and teachers
75-84	The group is able to convey explanations logically and smoothly and can answer questions from fellow students and teachers, but cannot manage time well
65-74	The group is able to convey explanations smoothly but is unable to convey the logic of their reasoning
55-64	The group is less able to convey explanations smoothly and on time and is less able to convey the logic of their reasoning
<55	

B. Criteria of Essay Score

Nilai	Answers Quality
100	The answer is very precise, all the definitions and main components are complete
76-99	The answer is quite precise, the meaning and the main components are almost complete
51-75	Inaccurate answers, incomplete understanding and main components
26-50	The answer is very inaccurate, the meaning and the main components are very incomplete
<25	Wrong answers

